

A NOVEL SYNTHESIS OF THE BASIC CARBON FRAMEWORK OF FREDERICAMYCIN A.
PROMISING ROUTES FOR THE SPIRO CHIRAL CENTER CONSTRUCTION
OF THE CD-RING SYSTEM

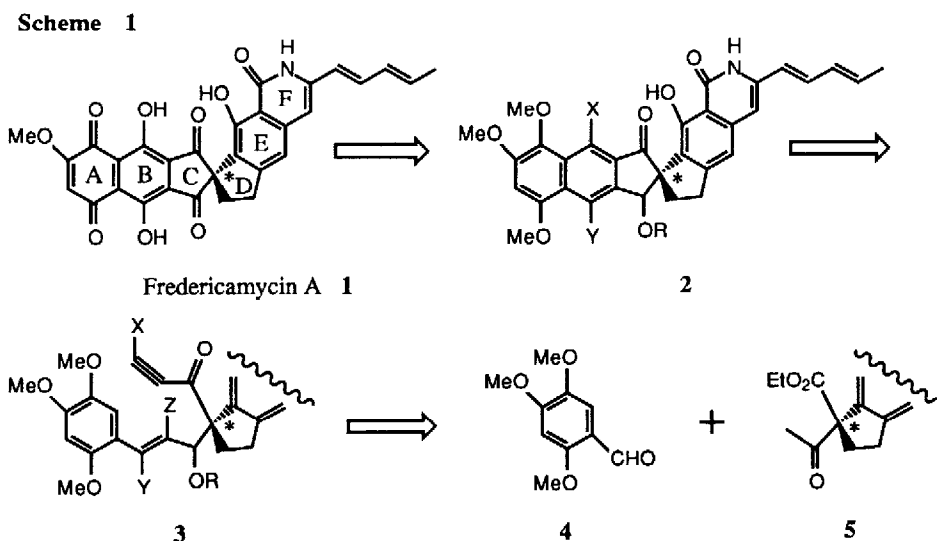
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Abstract: The Intramolecular Diels-Alder reactions of the dienyne readily obtainable from the aromatic aldehydes and ethyl 1-acetylcyclopentane-1-carboxylate, were found to proceed in a highly regioselective manner with formation of the novel title compounds as sole products.

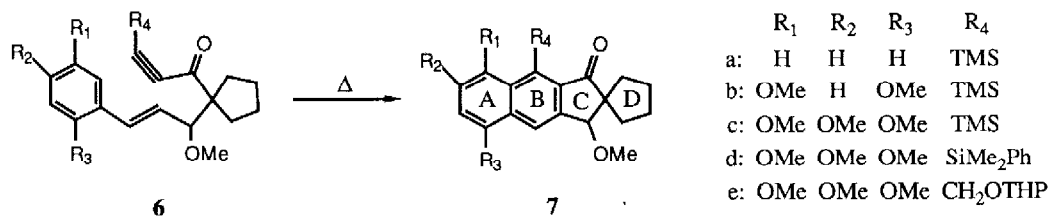
Fredericamycin A (**1**) has been distinguished as an unusually attractive target for total synthesis because of its prominent anticancer activity and characteristic spiro structure.¹⁾ While the initial structural elucidation of **1** reported by Pandey *et al.* had only concerned with its relative configuration,^{1b)} it was recently uncovered that **1** could be produced by *Streptomyces* species in an optically active form.²⁾ However, the absolute stereochemistry of naturally occurring **1** has not been determined.

Numerous synthetic approaches to **1** have hitherto been reported³⁾ and Kelly *et al.* has recently completed the first total synthesis of **1** in a racemic modification by employing elegant synthetic methodology.⁴⁾ We became interested in the unique structure of **1** involving a single chiral center at the spiro junction of the CD-ring system, and started the program directed at the total synthesis of optically active **1** of definite absolute configuration. However, all the synthetic methods so far developed,^{3a-g,i-k,n,p,q,4)} turned out to be not applicable to this



purpose because they have no potential for constructing an asymmetric center at the spiro junction of the CD-ring system in an enantioselective manner.

As a method for preparing **1** in an optically active form, the novel synthetic strategy pictured in **Scheme 1** was designed in which the regioselective intramolecular Diels-Alder reaction is ingeniously employed. Thus, when the intramolecular Diels-Alder reaction of the dienyne (**3**) obtainable from 2,4,5-trimethoxybenzaldehyde (**4**) and the optically active ethyl 1-acetylcyclopentane-1-carboxylate derivative (**5**), undergoes in a highly regioselective manner, the optically active hexacyclic compound (**2**) in which the asymmetric center is involved at the desired position can be produced after concomitant aromatization of the initially formed B-ring. Subsequent chemical manipulations of functionalities can give rise to optically active **1** from **2**. According to this synthetic route, it may be also possible to establish the absolute configuration of naturally occurring **1** by use of optically active **5** of the known absolute stereochemistry.

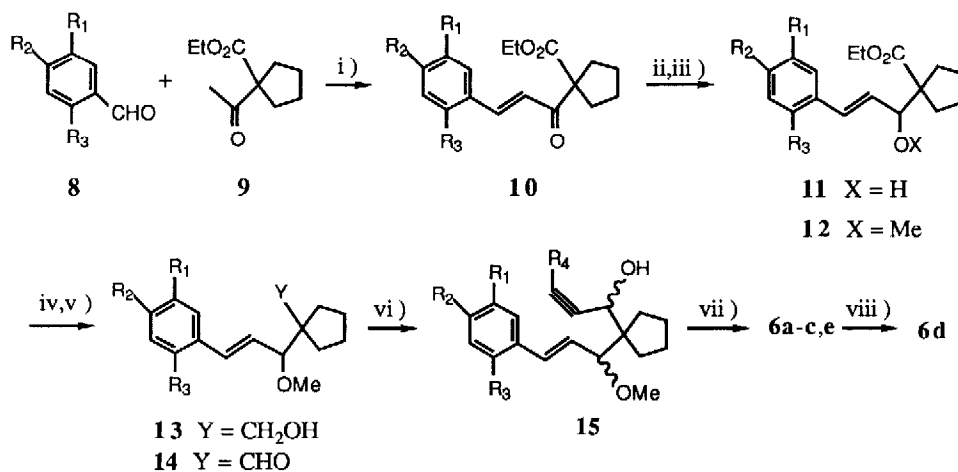


In order to explore feasibility of the designed synthetic strategy, the intramolecular Diels-Alder reactions of the dienyynes (**6a-e**), the model compounds of **3**, were first studied. We wish to report here that the intramolecular Diels-Alder reactions of **6b-e** proceed in a highly regioselective manner with exclusive formation of the 2,2-tetramethylene-benz[*f*]indane derivatives (**7b-e**) corresponding to **2**, as sole products.

The requisite dienyynes (**6a-e**) could be readily prepared following the reaction steps shown in **Scheme 2**. Thus, aldol condensations of the aromatic aldehydes (**8a-c**) with ethyl 1-acetylcyclopentane-1-carboxylate (**9**)⁵ followed by reductions of the enones (**10a-c**)⁶ and methylations of the secondary alcohols (**11a-c**), afforded the methyl ethers (**12a-c**). These were derived to the aldehydes (**14a-c**) by way of the primary alcohols (**13a-c**) by sequential reductions and oxidations. Additions of the lithium acetylides to **14a-c** followed by oxidations gave rise to **6a-c,e**. Exchange of the trimethylsilyl group of **6c** with a dimethylphenylsilyl group readily produced **6d**.

With **6a-e** in hand, the crucial intramolecular Diels-Alder reaction was first attempted using **6a**. Thus, when a solution of **6a** in *o*-dichlorobenzene was heated in an evacuated sealed tube, **7a** could be produced in 96% yield after concomitant oxidation of the initial addition product during the addition reaction. Being encouraged by this result, the intramolecular Diels-Alder reactions were next examined employing **6b-e** which bear the aromatic rings substituted with two or three methoxy groups to explore the regioselectivity of addition reaction. As summarized in **Table 1**, the reactions were found to proceed in a highly regioselective manner, affording **7b-e** in good to excellent yields. The structure of **7a-e** could be readily determined by their spectral data.⁷⁾

Scheme 2



i) LDA / THF, then **8**, a: 40% ; b: 81% ; c: 74%. ii) NaBH₄-CeCl₃·7H₂O / MeOH for **10a,c**, DIBAL-H / for **10b**, a: 95% ; b: 56% ; c: 97%. iii) NaH / DMF, then MeI for **11a,c**, MeI / Ag₂O for **11b**, a: 93% ; b: 69% ; c: 88%. iv) DIBAL-H / THF, a: 99% ; b: 61% ; c: 93%. v) SO₃·Py / DMSO / Et₃N for **13a,c**, PCC / CH₂Cl₂ for **13b**, a: 95% ; b: 60% ; c: 96%. vi) LiC≡CTMS for **14a-c**, LiC≡CCH₂O^tHP / Et₂O for **14e**, a: 98%, b: 82% ; c: 91% ; e: 72%. vii) MnO₂ / CH₂Cl₂ for **15a,c,e**, PCC / CH₂Cl₂ for **15b**, a: 96% ; b: 66% ; c: 77% ; e: 87%, viii) (1) ^tBu₄NF / THF (2) ^tBuLi / THF, then Me₂PhSiCl for **6c** : 43%.

Table 1 Results of the Intramolecular Diels-Alder Reactions of the Dienynes (**6a-e**)^{a)}

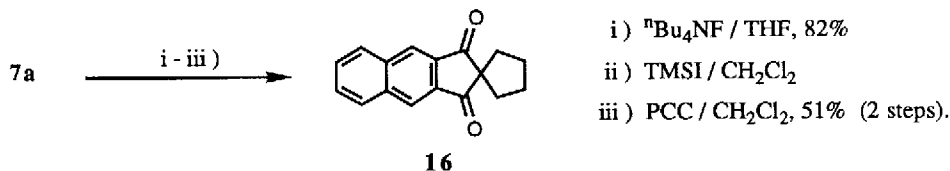
Run	6	Reaction Temp.(°C)	Condition Time (h)	Yields of 7 (%)
1	a	200	3	96
2	b	170	2	100
3	c	180	2	100
4	d	200	2	68 ^{b)}
5	e	200	2	73 ^{b)}

a) All reactions were carried out in *o*-dichlorobenzene.

b) Not optimized.

In order to firmly establish the assigned structures of **7a-e**, the simplest addition product (**7a**) was derived to known 2,2-tetramethylene-benz[*f*]indane-1,3-dione (**16**) previously synthesized by Bach *et al.*^{3g)} As shown in **Scheme 3**, desilylation of **7a** followed by cleavage of the methyl ether⁸⁾ and oxidation of the resulting secondary alcohol, smoothly produced **16**, mp. 131-132°C. The dione (**16**) was rigorously identified with an authentic sample, mp 128 - 129°C, by measuring the mixed melting point, mmp 130 - 131°C and comparing their

Scheme 3



spectral data.

As described above, the intramolecular Diels-Alder reactions of **6b-e** was found to proceed in a highly regioselective manner to afford good to excellent yields of **7b-e**. Application of this novel methodology to the synthesis of optically active **1** of the definite absolute configuration is in progress.

Acknowledgement: The authors are indebted to Professor R. D. Bach, Wayne State University, for providing us with an authentic sample of **16** and copies of its ^1H - and ^{13}C -NMR spectra.

References and Notes

- 1) a) R. C. Pandey, M. W. Toussaint, R. M. Stroshane, C. C. Kalita, A. A. Aszalos, A. L. Garretson, T. T. Wei, K. M. Byrne, R. F. Geoghegan, Jr., and R. J. White, *J. Antibiot.*, **34**, 1389 (1981). b) R. Misra, R. C. Pandey, and J. V. Silverton, *J. Am. Chem. Soc.*, **104**, 4478 (1982).
- 2) R. Misra, R. C. Pandey, B. D. Hilton, P. P. Poller, and J. V. Silverton, *J. Antibiot.*, **40**, 786 (1987).
- 3) a) A. V. R. Rao, D. R. Reddy, and V. H. Deshpande, *J. Chem. Soc. Chem. Commun.*, 1119 (1984). b) K. A. Parker, K. A. Koziski, and G. Breault, *Tetrahedron Lett.*, **26**, 2181 (1985). c) A. S. Kende, F. H. Ebetino, and T. Ohta, *Ibid.*, **26**, 3063 (1985). d) G. Eck, M. Julia, B. Pfeiffer, and C. Rolando, *Ibid.*, **26**, 4723 (1985). e) *Idem.*, *Ibid.*, **26**, 4725 (1985). f) M. Braun and R. Veith, *Ibid.*, **27**, 179 (1986). g) R. D. Bach and R. C. Klix, *Ibid.*, **27**, 1983 (1986). h) K. A. Parker and G. A. Breault, *Ibid.*, **27**, 3835 (1986). i) S. M. Bennett, and D. L. J. Clive, *J. Chem. Soc. Chem. Commun.*, 878 (1986). j) R. D. Bach and R. C. Klix, *J. Org. Chem.*, **51**, 749 (1986). k) M. A. Cinfolini and M. E. Browne, *Tetrahedron Lett.*, **28**, 171 (1987). l) A. V. R. Rao, D. R. Reddy, G. S. Annapurna, and V. H. Deshpande, *Ibid.*, **28**, 451,455 (1987). m) A. V. R. Rao, N. Sreenivasan, D. R. Reddy, and V. H. Deshpande, *Ibid.*, **28**, 455 (1987). n) G. Mehta and D. Subrahmanyam, *Ibid.*, **28**, 479 (1987). o) A. V. R. Rao and D. R. Reddy, *J. Chem Soc. Chem. Commun.*, 574 (1987). p) D. L. J. Clive, A. G. Angoh, and S. M. Bennett, *J. Org. Chem.*, **52**, 1339 (1987). q) S. N. Naik, B. Pandey, and N. R. Ayyangar, *Synth. Commun.*, **18**, 633 (1988).
- 4) T.R. Kelly, N. Ohashi, R. J. Armstrong-Chong, and S. H. Bell, *J. Am. Chem. Soc.*, **108**, 7100 (1986).
- 5) L. J. Goldsworthy, *J. Chem. Soc.*, 377 (1934).
- 6) J. L. Luche, *J. Am. Chem. Soc.*, **100**, 2226 (1978).
- 7) Representative IR and ^1H -NMR data of **7a-e** are as follows. **7a**: IR (CHCl_3) 1715 cm^{-1} ; ^1H -NMR (CDCl_3) 0.52 (9H, s, SiMe_3), 1.68-2.19 (8H, m, aliphatic H), 3.48 (3H, s, OMe), 4.59 (1H, s, CHOMe), 7.42-8.57 (5H, m, aromatic H). **7b**: ^1H -NMR (CDCl_3) 0.29 (9H, s, SiMe_3), 1.35-2.43 (8H, m, aliphatic H), 3.43 (3H, s, CHOMe), 3.85 and 3.98 (each 3H, each s, OMe x 2), 4.57 (1H, s, CHOMe), 6.69 and 6.86 (each 1H, each d, J 9.0 Hz, aromatic H x 2), 8.38 (1H, s, aromatic H). **7c**: ^1H -NMR (CDCl_3) 0.31 (9H, s, SiMe_3), 1.58-2.20 (8H, m, aliphatic H), 3.44 (3H, s, CHOMe), 3.61, 4.02 and 4.06 (each 3H, each s, OMe x 3), 4.54 (1H, s, CHOMe), 6.73 and 8.32 (each 1H, each s, aromatic H x 2). **7d**: ^1H -NMR (CDCl_3) 0.51 and 0.53 (each 3H, each s, Me x 2), 1.40-2.22 (8H, m, aliphatic H), 3.20 (3H, s, CHOMe), 3.45, 3.89 and 4.01 (each 3H, each s, OMe x 3), 4.52 (1H, s, CHOMe), 6.70 (1H, s, aromatic H), 7.13-7.69 (5H, m, aromatic H), 8.37 (1H, s, aromatic H). **7e**: ^1H -NMR (CDCl_3) 1.14-2.21 (14H, m, aliphatic H), 3.43 and 3.44 (each 1.5H, each s, OMe), 3.92 (3H, s, OMe), 4.01 (6H, s, OMe x 2), 4.48 (1H, s, CHOMe), 4.92-5.08 (1H, m, -OCHO-), 5.47 and 5.97 (each 1H, each dd, J 9.0 and 12.0 Hz, ArCH_2O -), 6.80 and 8.40 (each 1H, each s, aromatic H).
- 8) M. E. Jung and M. A. Lyster, *J. Org. Chem.*, **42**, 3761 (1977).

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